

Photocatalytic performance of carbon modified BiVO₄ for degradation of phenol under visible light

Minli Niu¹, Rongshu Zhu^{1,2,3*}, Fei Tian¹, Kelin Song¹

¹Environmental Science and Engineering Research Center Harbin Institute of Technology Shenzhen Graduate School, Shenzhen 518055, P. R.China.

²Shenzhen Key Laboratory of Water Resource Utilization and Environmental Pollution Control, Shenzhen 518055, P. R.China.

³Public Platform for Technological Service in Urban Waste Reuse and Energy Regeneration, Shenzhen 518055, P. R.China.

*corresponding author: rszhu@hitsz.edu.cn

Introduction

BiVO₄ is an excellent visible-light photocatalyst that is attributed to its narrow band gap, hemical stability and nontoxicity. However, the photocatalytic activity of the pure BiVO₄ still needs improvement due to the high recombination rate of photogenerated e⁻-h⁺ pairs. Recently, Wang et al. [1] synthesized carbon modified TiO₂ nanotube array photocatalysts by sucrose graphitization, and found that these photocatalysts showed the enhanced photoelectric catalytic activity due to the high migration efficiency of photoinduced electrons at the graphite-like carbon/TiO₂ interface. Therefore, It is worth researching how to reduce the recombination rate of photogenerated e⁻-h⁺ pairs at the carbon/BiVO₄ interface. Moreover, in order to study the effect of the carbon precursors on the photocatalytic activity better, sucrose and glucose were used in the preparation of catalyst. The photocatalytic activities were evaluated by phenol degradation under visible-light irradiation.

Materials and Methods

The catalysts was prepared by a hydrothermal method firstly and then calcination [2]. Sucrose and glucose were used as the carbon precursor. In the photocatalytic experiments, a 350 W Xe lamp was used as the light source and the UV part of the light was removed by a cut-off filter ($\lambda > 420\text{nm}$). In all experiments, 200 mL 5M phenol solution and 0.2 g of catalyst was added into the reaction cell. Adsorption was conducted in the dark for 30 min before the lighting started. The temperature for all the photocatalytic reactions was kept at $25 \pm 1^\circ\text{C}$. Illumination time was 3h and detection instrument was HPLC.

Results and Discussion

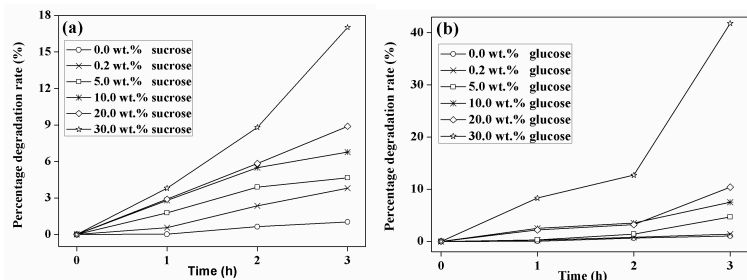


Fig. 1. The photocatalytic activities of carbon modified BiVO₄ for the degradation of phenol, (a) sucrose and (b) glucose

The photocatalytic activities of carbon modified BiVO₄ for the degradation of phenol are shown in Fig. 1. It can be seen from Fig. 1 that the phenol degradation rate increases with the increase of carbon precursors amount for both sucrose and glucose. The result indicates that the modification of carbon increase the photocatalysis of BiVO₄.

The effects of carbon precursors on the degradation of phenol are shown in Fig. 2. As can be seen, the catalysts with glucose and sucrose as the precursors have similar photocatalytic activity when precursor amount is less than 20 wt.%. However, the former has higher photocatalytic activity when precursor amount is greater than 20 wt.%. The result indicates that the choice of the carbon precursor had a great influence on the photocatalytic activity of BiVO₄.

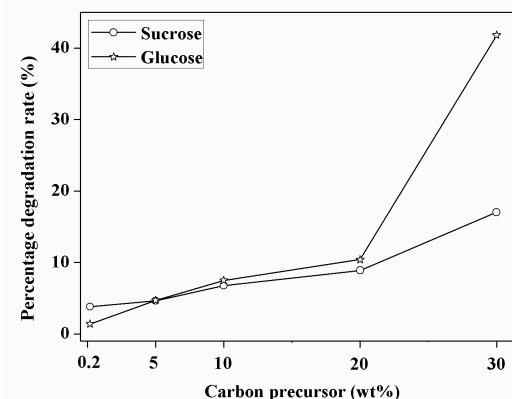


Fig. 2. The effects of carbon precursors on the degradation of phenol

Significance

Carbon modified BiVO₄ can enhance photocatalytic activity in degradation of phenol under visible light. The choice of the carbon precursors had a great influence on the photocatalytic activity of BiVO₄. The degradation rate increases with the increase of carbon precursors amount. Catalyst has much higher photocatalytic activity when the carbon precursors amount is greater than 20 wt.%.

Acknowledgements

All the authors gratefully acknowledge support from the Special fund for the development of strategic and new industry in Shenzhen (No. JCYJ20120613114951217), the Fund for the Research and Development of Science and Technology in Shenzhen (No. CXZZ20130516145955144) and the National High Technology Research and Development Program of China (2012ZX07206-002).

References

- Wang, Y. J.; Lin, J.; Zong, R. L.; He, J.; Zhu, Y. F. *Journal of Molecular Catalysis A: Chemical* 2011, 349, 13-19.
- Zhao, W. R.; Wang, Y.; Yang, Y.; Tang, J.; Yamg, Y. N. *Applied Catalysis B: Environmental* 2012, 115-116, 90-99.